Ley de los Gases

Ley de Charles (P=cte)

$$\frac{V_1}{T_1} = \frac{V_2}{T_2}$$

$$\frac{V}{T} = \text{constante}$$

Ley de los Gases Ideales

$$\frac{P_1 \cdot V_1}{T_1} = \frac{P_2 \cdot V_2}{T_2}$$

$$\frac{P \cdot V}{T} = \text{constante}$$

Ley de Boyle - Mariotte (T=cte)

$$P_1 \cdot V_1 = P_2 \cdot V_2$$

$$P \cdot V = \text{constante}$$

Ley de Gay Lussac (V=cte)

$$\frac{P_1}{T_1} = \frac{P_2}{T_2}$$

$$\frac{P}{T} = \text{constante}$$